



ID Card

Diammonium hexachloropalladate

Version 18 July 2023

Notes:

- This ID card is used to support the substance sameness discussions in SIEFs and to describe the substance to the best of the SIEF members' knowledge.
- It also aims at grouping communications relevant to the request of available data or information, the approval of the proposed Lead Registrant and the registration strategy with the SIEF.
- It is the responsibility of each individual registrant to identify their substance and to report company-specific identity in their Registration Dossier (section 1.2 of IUCLID).

DISCLAIMER

All data and information contained in this document shall be treated by the receiving party (i) in full confidence with the adequate respect of any confidential and/or proprietary nature of such information and (ii) only in the framework of the purpose of agreeing on substance sameness, Lead Registrant and overall REACH Strategy for the concerned Substance under REACH (the 'Purpose').

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1. Identification of the substance

Table 1. Identification of the substance

	Original (in EC inventory)
Name	Diammonium hexachloropalladate
EC number	242-854-9
CAS number	19168-23-1
Description	Not available
Composition type	Mono-constituent substance

2. Synonyms and other identifiers of the substance

Table 2. Synonyms and other identifiers of the substance

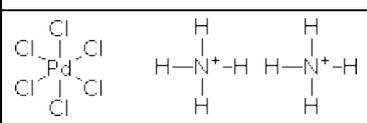
IUPAC name	Diammonium hexachloropalladate(2-)
CAS name	Palladate(2-), hexachloro-, ammonium (1:2), (OC-6-11)-
Abbreviations	None
Other commercial or international names	Red Salt Ammonium chloropalladate Ammonium Hexachloro Palladate (IV) Ammonium hexachloropalladate Hexachloropalladate(2-) diammonium
Other identity codes	None

3. Substances (with core identifiers) also falling under this substance (with justification)

None

4. Information related to molecular and structural formula of the substance

Table 3. Information related to molecular and structural formula of the substance

Molecular formula	H8Cl ₆ N ₂ Pd
Structural formula	
Smiles notation	Cl[Pd-2](Cl)(Cl)(Cl)(Cl)Cl.[NH4+].[NH4+]
Optical activity	Not available
Typical ratio of (stereo) isomers	Not available
Molecular Weight / Molecular Weight range	355,21 g/mol

5. Typical composition of the substance

Table 4. Typical composition

	Name	Symbol / Formula	Min & Max concentrations (%)[§]	Typical concentration (%)^{§§}
Main constituent(s)*	Diammonium hexachloropalladate	H8Cl ₆ N ₂ Pd	99,5 - 100 [§]	> 99,5
Impurity(ies)#	Several minor (especially metallic) impurities which do not affect the classification of the substance because of their non-hazardous nature or because they do not exceed the classification cut-off limits in the substance	e.g. Ag, Au, Cu, Ir, Pb, Pt, Rh, Ru	0 - 0,5	< 0,5

* ≥ 80 % (w/w) for mono-constituent substances; ≥ 10 % (w/w) and < 80 % (w/w) for multi-constituent substances.

An impurity is an unintended constituent present in a substance, as produced. It may originate from the starting materials or be the result of secondary or incomplete reactions during the production process. While impurities are present in the final substance, they were not intentionally added.

§ Concentration ranges define the substance sameness criteria agreed by all Consortium Members in preparation of the communication with other SIEF members.

§§ Typical concentration refers to the representative sample used for testing.

§ Corresponds to 29,8 - 30 % Pd.

The composition given above is typical and should therefore represent the majority of Diammonium hexachloropalladate as manufactured and/or imported in the EEA market. Diammonium hexachloropalladate containing less than 99,5 % Diammonium hexachloropalladate may still be considered to be the same for the purpose of registration under REACH and may be referred to as impure

Diammonium hexachloropalladate to distinguish it from the typically pure Diammonium hexachloropalladate.

6. Information on appearance, physical state and properties of the substance

Table 5. Appearance / physical state / properties of the solid substance

Physical state	Solid
Physical form*	Crystalline form
Appearance	Brown/red lumps and powder
Particle size**	Fine powder to coarse powder
Does the solid hydrolyse?#	No
Is the solid hygroscopic?§	No

* Crystalline form: solid material whose constituent atoms, molecules, or ions are arranged in an ordered pattern extending in all three spatial dimensions. Amorphous form: solid material whose constituent atoms, molecules, or ions are randomly arranged.

** Nanoform: particles in the size range 1 - 100 nm (for full definition of a nanomaterial, see <http://ec.europa.eu/environment/chemicals/nanotech/index.htm#definition>). Fine powder: particles in the size range 100 – 2.500 nm. Coarse powder: particles in the size range 2.500 nm – 1 mm. Massive object: particles in the size range > 1 mm.

Hydrolysis: decomposition (cleavage of chemical bonds) by the addition of water.

§ Hygroscopic substance: readily attracts moisture from its surroundings in open air, through either absorption or adsorption. Cf. also water/moisture content in Table 4.

7. Analytical data

Annex VI of REACH requires the registrant to describe the analytical methods and/or to provide the bibliographical references for the methods used for identification of the substance and, where appropriate, for the identification of impurities and additives. This information should be sufficient to allow the methods to be reproduced.

Table 6. Analytical methods for identification of the substance

Parameter / Method	Recommended for substance identification and sameness check	Applicable	Not applicable or not recommended
Elemental analysis			
ICP (ICP-MS or ICP-OES)	X		
Atomic absorption spectroscopy (AAS)			
Glow discharge mass spectrometry (GDMS)			
Molecular analysis			
Infrared (IR) spectroscopy	X		
Raman spectroscopy			X
Mineralogical analysis			
X-Ray Fluorescence (XRF)		X	



X-Ray Diffraction (XRD)	X		
Morphology and particle sizing			
Electron microscopy (SEM, TEM, REM)*#			
Laser diffraction*#	X		
Particle size by other means (e.g. sieve analysis)#			
Surface area by N-BET*#	X		
Other			

* Analytical techniques particularly (but not exclusively) relevant for nanomaterials.

The choice of the technique for particle size depends on the size of the material as manufactured/imported/placed on the market/used.

8. Lead Registrant

Johnson Matthey (United-Kingdom) volunteers to be the Lead Registrant for Diammonium hexachloropalladate. The EPMF will provide support to the Lead Registrant as laid down in the EPMF Agreement.

9. Scope of the Registration Dossier

The uses included in this Registration Dossier are listed on the [EPMF website](#).

10. Analytical reference information

Below the results of IR analysis of a reference sample used for testing.

Instrument: Bruker Tensor 37

Measurement head: Diamond ATR

Resolution: 2 cm⁻¹

Sample scans: 16

Background scans: 16

Atmospheric compensation: Off

Result spectrum: Absorbance

Source: MIR

Beam splitter: KBr

Aperture: 5 mm

Detector: DLaTGS

Detector temperature: Ambient

Scanner velocity: 10 kHz

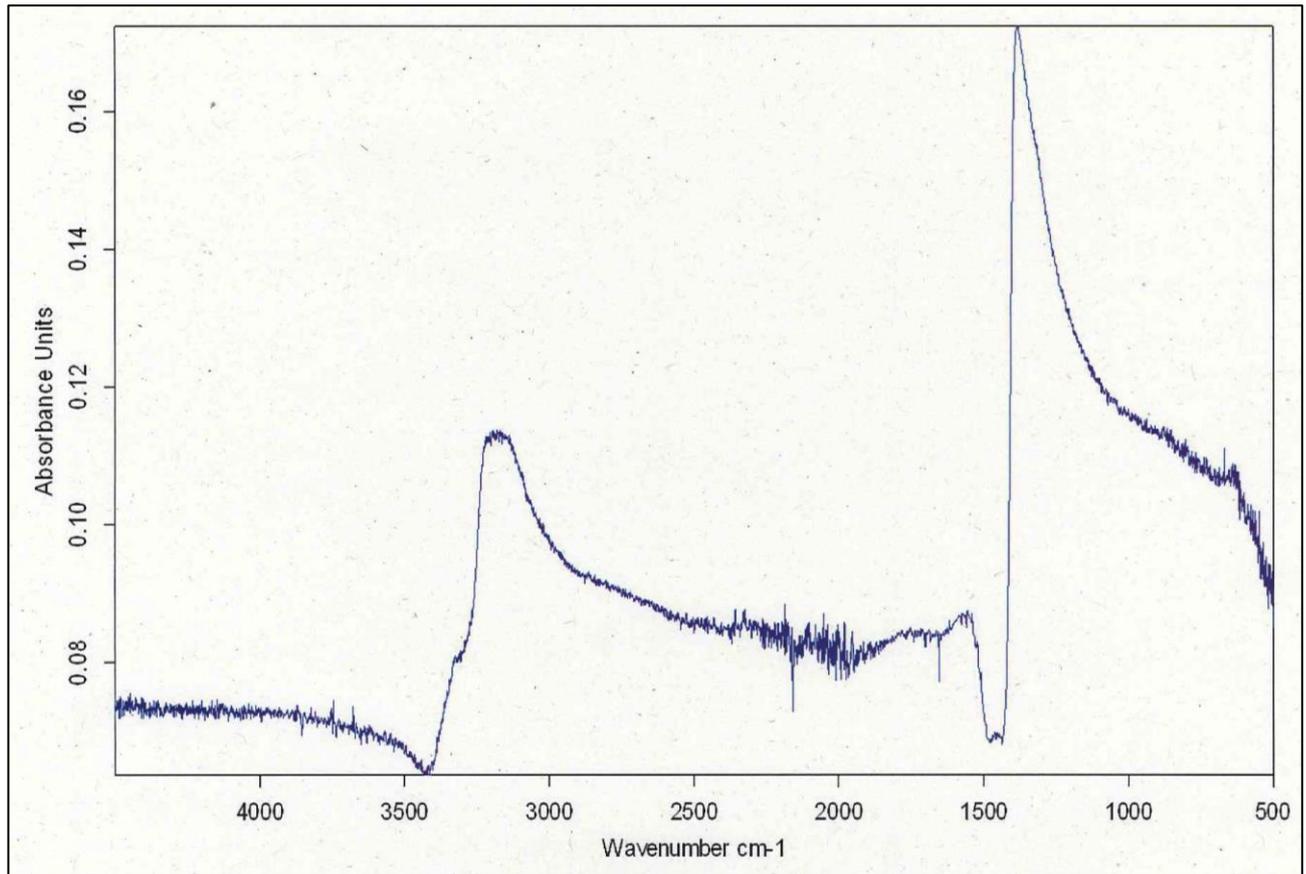


Figure 1. Typical infrared spectrum of Diammonium hexachloropalladate.